

Unexpected formation of a chiral δ -lactone by reduction of the 1,3-dicarbonylic cyclopentanoid derivatives

Luciane F. de Oliveira and Valentim E. U. Costa*

Departamento de Química Orgânica, Instituto de Química, Universidade Federal do Rio Grande do Sul, Av. Bento Gonçalves, 9500, 91501-970 Porto Alegre RS, Brazil

Received 14 February 2006; revised 7 March 2006; accepted 9 March 2006

Abstract—We have described the synthesis of highly functionalized chiral cyclopentanoids, which are important building units for synthesis of biological active compounds. The (–)- or (+)-7,7-dimethoxy-1,4,5,6-tetrachlorobicyclo[2.2.1]hept-5-en-2-endo-yl acetate, obtained from the enzyme catalyzed transesterification of the racemate, was converted to α -diketone chiral. The α -diketone was treated with $\text{H}_2\text{O}_2/\text{NaOH}$ and esterified with CH_2N_2 to furnish a mixture of the compounds (+)- or (–)-**10** and (+)- or (–)-**11**. The reduction of the (+)- or (–)-**10** and/or (+)- or (–)-**11** with $\text{BH}_3\cdot\text{THF}$ furnished the lactone (+)- or (–)-**13** with excellent yield. The α -diketone was reduced with indium metal in the presence of NH_4Cl furnishing the acyloin (+)-**14** in 67% of yield. The treatment of acyloin (+)-**14** with $\text{Pb}(\text{OAc})_4$ furnished the aldehyde (+)-**15** with 80% of yield. The reduction of the aldehyde (+)-**15** with NaBH_4 has again produced the lactone (+)-**13**.

© 2006 Elsevier Ltd. All rights reserved.

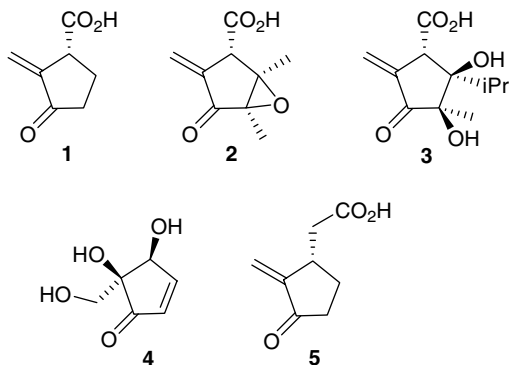
Cyclopentanoids products are a biologically important class of compounds, constituting structural part of prostaglandins¹ and there are several examples of this class of compounds acting as antitumor agents.² Since many natural products and synthetic materials of biological importance have cyclopentyl skeletons, in the last few years there has been intense development of synthetic methods to construct such moieties.³ Aminocyclopentanoids have been reported to be potent glycosidase inhib-

itors.⁴ Some examples of antitumor agents² are the sarkomycin **1**, the methylenomycin **2**, the xanthocidin **3**, the pentenomycin **4**, and the homosarkomycin **5**.

Several methodologies of asymmetric synthesis of the chiral building blocks with cyclopentane ring have been developed.⁵ Khan et al.⁶ described an efficient synthetic methodology employing catalytic $\text{RuCl}_3\cdot 3\text{H}_2\text{O}$ and NaIO_4 as stoichiometric cooxidants to get α -diketones from the vicinal dihaloalkenes. α -Diketones were cleaved using $\text{Pb}(\text{OAc})_4$ or alkaline H_2O_2 to give cyclopentanoids derivatives in their racemic form.

A convenient method for obtaining enantiopure materials is catalyzed acylation.⁷ The enzyme catalyzed transesterification using vinyl acetate was applied for resolution of 7,7-disubstituted-1,4,5,6-tetrachlorobicyclo[2.2.1]hept-5-en-2-ols using lipase from *Candida rugosa*.⁸

In the last years, enzymatic resolution has been used by our group to obtain enantiopure alcohols and their derivatives,⁹ and, recently, we described an efficient synthesis of enantiopure 1,2-¹⁰ and 1,3-amino alcohols¹¹ from the 7,7-dimethoxy-1,4,5,6-tetrachlorobicyclo[2.2.1]hept-5-en-2-endo-ol (**6**), which were obtained by enzymatic procedure using the lipase from *Candida rugosa*



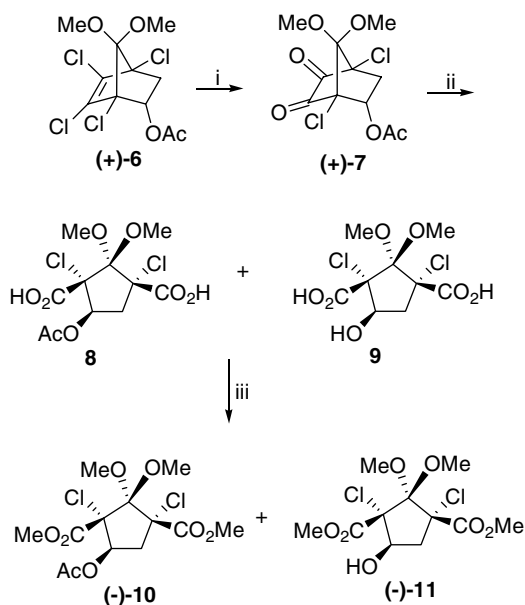
* Corresponding author. Tel.: +55 051 33166300; fax: +55 051 33167304; e-mail: valentim@iq.ufrgs.br

in vinyl acetate. We now report the asymmetric synthesis of the interesting chiral cyclopentanoids synthons **8** and **9** from the enantiopure acetate (+)-**6**.

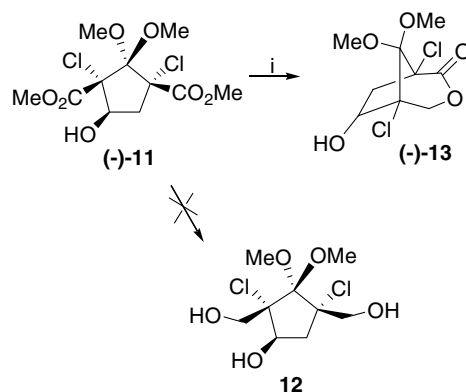
(±)-7,7-Dimethoxy-1,4,5,6-tetrachlorobicyclo[2.2.1]hept-5-en-2-endo-ol, obtained from the Diels–Alder reaction of 1,2,3,4-tetrachloro-5,5-dimethoxycyclopentadiene and vinyl acetate followed by acid hydrolysis (sulfuric acid/methanol), was subjected to transesterification⁸ with vinyl acetate and lipase from *Candida rugosa*, over 7 days.

The enantiomeric purity of the products was determined by chiral gas chromatography (alcohol: ee 98%; acetate: ee 99%). The alcohol was acetylated (acetic anhydride, DMAP) to furnish the corresponding acetate (+)-**6**.

Using the Khan^{6b} procedure, the acetate (+)-**6** was subjected to an oxidative cleavage reaction using a catalytic amount of RuCl₃ anhydrous in the presence of NaIO₄ to give α-diketone (+)-**7** in excellent yield (Scheme 1). We subsequently examined the H₂O₂/NaOH mediated cleavage reaction of α-diketone (+)-**7**. When this reaction was employed, a mixture of **8** and **9** was obtained. This mixture was esterified to the dimethyl ester (–)-**10** and (–)-**11**, purified by flash chromatography furnishing acetate (–)-**10**¹² and alcohol (–)-**11**¹³ in a ratio of 4:1 in quantitative yield. However, the reduction of this mixture with LiAlH₄ was unsuccessful due to its lower yield. The reduction of the alcohol (–)-**11**, protected with THP, with LiAlH₄ also gave lower yield. Therefore, with a purpose to obtain the triol **12**, the compound (–)-**11** was subjected to reduction with BH₃·THF at room temperature.¹⁴ Surprisingly, this reduction furnished just the lactone (–)-**13**, instead of the alcohol **12** (Scheme 2). Carboxylic acids can be reduced with boranes more easily than with esters, so, we subject the mixture of **8** and **9**



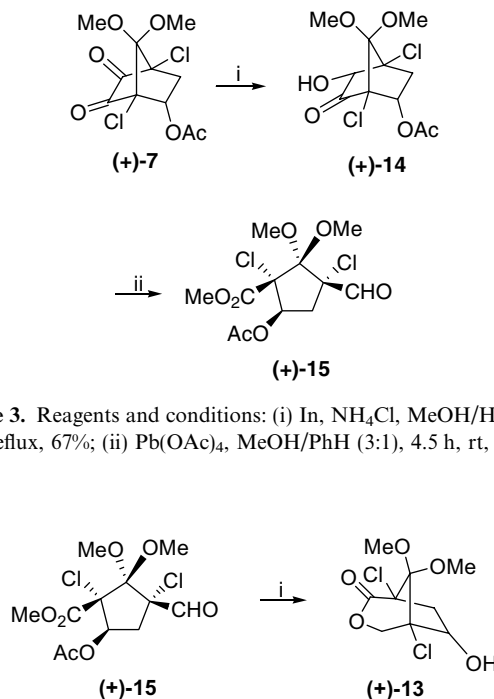
Scheme 1. Reagents and conditions: (i) RuCl₃, NaIO₄, MeCN/H₂O (6:1), rt, 10 min, 99%; (ii) H₂O₂/NaOH, MeOH, rt, 2 h; (iii) CH₂N₂, Et₂O, 0 °C, 15 min, 90%.



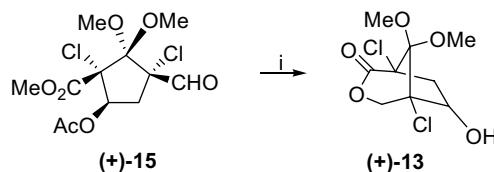
Scheme 2. Reagents and conditions: (i) BH₃·THF, 24 h, rt, 90%.

to a reduction with BH₃·THF or BH₃·SMe₂, in the same reaction conditions; however, always the lactone (–)-**13**, was obtained with high yields.

Since the attempted reduction of **8**, **9**, **10**, or **11** furnish always of the lactone **13**, we have decided to make use of the Khan¹⁵ methodology to obtain cyclopentane carboxaldehydes. The α-diketone (+)-**7** was reduced with indium metal in MeOH/H₂O (4:1) in the presence of NH₄Cl furnishing the acyloin (+)-**14** in 67% of yield.¹⁶ It is important to use dry MeOH even though water is also used as a cosolvent, because it is crucial for the solvent ratio MeOH/H₂O (4:1). The treatment of acyloin (+)-**14** with Pb(OAc)₄ in MeOH/PhH (3:1) furnished the aldehyde¹⁷ (+)-**15** with 80% of yield (Scheme 3). Surprisingly, the reduction of the aldehydes (+)-**15** with NaBH₄ in MeOH has again produced the same lactone (+)-**13** (Scheme 4).



Scheme 3. Reagents and conditions: (i) In, NH₄Cl, MeOH/H₂O (4:1), 23 h, reflux, 67%; (ii) Pb(OAc)₄, MeOH/PhH (3:1), 4.5 h, rt, 80%.



Scheme 4. Reagents and conditions: (i) NaBH₄, MeOH, 8 h, rt, 89%.

In conclusion, we have described the synthesis of chiral highly functionalized cyclopentanoids, which are potential building blocks in organic synthesis. The obtainment of the lactone from the α -chloro diester and α -chloro diacid was achieved with excellent yields.

Acknowledgments

Financial support from CNPq and FAPERGS is acknowledged. We thank also CNPq for fellowship to V.E.U.C. and CAPES, for scholarship to L.F.O. The authors are indebted to Amano Enzyme USA Co. Ltd, for providing the 'Amano' lipases used in this work. We also thank Professor Khan by experimental details of the procedure for reduction of α -diketone.

References and notes

- Kobayashi, Y.; Murugesu, M. G.; Nakano, M.; Takahisa, E.; Usmani, S. B.; Aina, T. *J. Org. Chem.* **2002**, *67*, 7110–7123.
- Martinez, A. G.; Vilar, E. T.; Fraile, A. G.; Cerero, S. M.; Osuna, S. O.; Maroto, B. L. *Tetrahedron Lett.* **2001**, *42*, 7795–7799.
- (a) Trost, B. M.; Pinkerton, A. B. *J. Org. Chem.* **2001**, *66*, 7714–7722, and references cited therein; (b) Hudlicky, T.; Price, J. D. *Chem. Rev.* **1989**, *89*, 1467–1486.
- Denmark, S. E.; Dixon, J. A. *J. Org. Chem.* **1998**, *63*, 6178–6195.
- Johnson, T. A.; Curtius, M. D.; Beak, P. *Org. Lett.* **2002**, *4*, 2747–2749, and references cited therein.
- (a) Khan, F. A.; Dash, J.; Sahu, N.; Sudheer, C. *J. Org. Chem.* **2002**, *67*, 3783–3787; (b) Khan, F. A.; Prabhudas, B.; Dash, J.; Sahu, N. *J. Am. Chem. Soc.* **2000**, *122*, 9558–9559; (c) Khan, F. A.; Satapathy, R.; Sudheer, Ch.; Rao, Ch. N. *Tetrahedron Lett.* **2005**, *46*, 7193–7196.
- Chen, C. S.; Shi, C. J. *Angew. Chem., Int. Ed. Engl.* **1989**, *28*, 695–707.
- Berger, B.; Rabiller, C. G.; Konigsberger, K.; Faber, K.; Gringl, I. *Tetrahedron: Asymmetry* **1990**, *1*, 541–546.
- (a) Costa, V. E. U.; Alifantes, J.; Martins, J. E. D. *Tetrahedron: Asymmetry* **1998**, *9*, 2579–2585; (b) Morisso, F. D. P.; Costa, V. E. U. *Tetrahedron: Asymmetry* **2001**, *12*, 2641–2647; (c) Alifantes, J.; Nichele, A. G.; Costa, V. E. U. *Tetrahedron: Asymmetry* **2002**, *13*, 2019–2024.
- Lapis, A. A. M.; Kreutz, O. C.; Pohlmann, A. R.; Costa, V. E. U. *Tetrahedron: Asymmetry* **2001**, *12*, 557–561.
- de Oliveira, L. F.; Costa, V. E. U. *Tetrahedron: Asymmetry* **2004**, *15*, 2583–2590.
- $[\alpha]_D^{20}$ –8 (c 5.0, AcOEt). IR (film): ν (cm⁻¹): 1746. ¹H RMN (300 MHz, CDCl₃): δ 2.01 (s, 3H), 2.53 (dd, J = 14.3, 6.4 Hz, 1H), 3.18 (s, 3H), 3.83 (dd, J = 14.3, 10.5 Hz, 1H), 3.61 (s, 3H), 3.76 (s, 3H), 3.78 (s, 3H), 5.56 (dd, J = 10.5, 6.4 Hz, 1H). ¹³C RMN (75 MHz, CDCl₃): δ 20.7, 41.5, 52.5, 53.2, 53.4, 53.5, 75.8, 77.7, 78.7, 108.3, 165.9, 167.5, 169.4.
- $[\alpha]_D^{20}$ –6 (c 2.2, AcOEt), mp = 78–80 °C. IR (film): ν (cm⁻¹): 3507 (OH), 1743 (C=O). ¹H RMN (300 MHz, CDCl₃): δ 2.61 (dd, J = 14.7, 8.2 Hz, 1H), 3.28 (dd, J = 14.7, 8.2 Hz, 1H), 3.31 (s, 3H), 3.58 (d, J = 9.1 Hz, OH), 3.73 (s, 3H), 3.84 (s, 3H), 3.85 (s, 3H), 4.65 (m, 1H). ¹³C RMN (75 MHz, CDCl₃): δ 44.2, 52.4, 53.4, 53.6, 53.7, 74.7, 76.2, 79.2, 109.5, 167.8, 168.3.
- General procedure for obtainment of the lactone: To a solution of the acetate (–)-**10** (250 mg, 0.67 mmol) in dry THF (5 mL) and under argon, BH₃·THF (5 mL, 5 mmol, 1 M) was slowly added. The mixture was stirred at room temperature for 24 h and then was added MeOH (50 mL). The solution was stirred for 1 h and the solvent evaporated under reduced pressure. The crude product was purified by flash chromatography (silica, cyclohex/AcOEt 85:15) yielding 164 mg (0.605 mmol, 90%) of (–)-**13** as a white solid. Mp = 195–197 °C, $[\alpha]_D^{20}$ –9 (c 1.5, AcOEt). IR (film): ν (cm⁻¹): 3406, 1738. ¹H RMN (300 MHz, CDCl₃): δ 2.34 (dd, J = 14.8, 4.4 Hz, 1H), 2.79 (dd, J = 14.8, 10.9 Hz, 1H), 3.63 (s, 3H), 3.75 (s, 3H), 4.25 (dd, J = 10.4, 1.1 Hz, 1H), 4.61 (d, J = 10.4 Hz, 1H), 4.66 (dd, J = 10.9, 4.4 Hz, 1H). ¹³C RMN (75 MHz, CDCl₃): δ 44.6 (CH₂), 52.4 (CH₃), 52.9 (CH₃), 66.9 (CH₂), 72.6 (C), 72.7 (C), 72.9 (CH), 101.2 (C), 167.8 (C=O). HRMS found: m/z = 271.0140; calcd for C₉H₁₂Cl₂O₅ [M]⁺: 271.0177.
- Khan, F. A.; Dash, J.; Sahu, N.; Gupta, S. *Org. Lett.* **2002**, *4*, 1015–1018.
- General procedure for reduction of α -diketone: A mixture of α -diketone (0.3 mmol), indium metal (0.6 mmol, cut into small pieces) and NH₄Cl (0.9 mmol) in dry MeOH (4 mL) and water (1 mL) was refluxed for 23 h. It is important to use very small pieces of indium metal, with very good stirring and most crucial is the solvent ratio MeOH/H₂O (4:1). The reaction mixture was quenched with 3 mL of 5% HCl and extracted with ethyl acetate. The crude product was purified by flash chromatography gave the acyloin with 67% of yield. Mp = 98–100 °C, $[\alpha]_D^{20}$ +10 (c 1.25, AcOEt); more details related to the spectroscopy data are available in the literature.¹⁵
- Aldehyde (+)-**15**: $[\alpha]_D^{20}$ +12 (c 1.02, AcOEt); more details related to the spectroscopy data are available in the literature.¹⁵